

UNCLASSIFIED

AD NUMBER

AD014839

CLASSIFICATION CHANGES

TO: **unclassified**

FROM: **confidential**

LIMITATION CHANGES

TO:

**Approved for public release, distribution
unlimited**

FROM:

**Distribution authorized to DoD and DoD
contractors only; Foreign Government
Information; 14 JUL 1953. Other requests
shall be referred to The British Embassy,
3100 Massachusetts Avenue, NW, Washington,
DC 20008.**

AUTHORITY

**DSTL, WO 189/705, 5 Mar 2009; DSTL, WO
189/705, 5 Mar 2009**

THIS PAGE IS UNCLASSIFIED

CONFIDENTIAL

AD 14839

*Reproduced
by the*

**ARMED SERVICES TECHNICAL INFORMATION AGENCY
ARLINGTON HALL STATION
ARLINGTON 12, VIRGINIA**



**CLASSIFICATION CHANGED
TO CONFIDENTIAL
FROM SECRET**

**PER AUTHORITY LISTED IN
ASTIA TAB NO. 60-3-4
DATE 1 SEP 1960**

CONFIDENTIAL

SECRET

P.T.P. 371

COPY No. 69

D. NO. 148
VITR
RECORDED

MINISTRY OF SUPPLY

DEPARTMENT OF CHEMICAL DEFENCE RESEARCH AND DEVELOPMENT

CHEMICAL DEFENCE EXPERIMENTAL ESTABLISHMENT

**AIR-WAY RESISTANCE CHANGES IN
MEN EXPOSED TO GB VAPOUR**

1. THIS INFORMATION IS DISCLOSED FOR OFFICIAL USE BY THE RECIPIENT GOVERNMENT. DISCLOSURE TO ANY OTHER GOVERNMENT OR RELEASE TO THE PRESS OR IN ANY OTHER WAY WOULD BE A BREACH OF HIS CONFIDENCE.
2. THE INFORMATION SHOULD BE PRESERVED AND BE PASSED ON TO GIVE THE SAME STANDARD OF SECURITY AS THAT MAINTAINED BY HER MAJESTY'S GOVERNMENT IN THE UNITED KINGDOM.
3. THE INFORMATION CONTAINED IN THIS REPORT SHOULD NOT BE CIRCULATED OR SHOWN TO ANY MEMBER OF PARLIAMENT WITHOUT THE PRIOR PERMISSION OF THE MINISTRY OF SUPPLY.

M. AINSWORTH AND J.W. EVELEIGH

MORTON TECHNICAL PAPER No. 371

C.D.E.E.
Porton.
Wiltshire

SECRET

S E C R E T

PORTON TECHNICAL PAPER NO: 371
COPY NO: 69
DATE: 14th July, 1953.

AIR-WAY RESISTANCE CHANGES IN MEN EXPOSED TO GB VAPOUR

by

M. Ainsworth and J.W. Eveleigh

SUMMARY

Changes in lung air-way resistance to flow in men exposed by inhalation to GB vapour have been estimated. Measurable increases resulted from the administration of doses larger than 1.0 microgram per kg.

In nearly every case, the increase was maximal immediately after gassing and then diminished fairly rapidly in an exponential manner. The degree of the effect appeared to depend on the dose and the concentration of the inhaled GB vapour.

(Sgd). H. Cullumbine,

Head, Physiology Section.

(Sgd). E.A. Perren,

Supt., Research Division.

ML/JWE/GC.

S E C R E T

S E C R E T

PORTON TECHNICAL PAPER NO: 371

COPY NO: 69

DATE: 14th July, 1953.

AIR-WAY RESISTANCE CHANGES IN MEN EXPOSED TO GB VAPOUR

by

M. Ainsworth and J.W. Eveleigh

INTRODUCTION

The failure of respiration in poisoning by the nerve gases is a result of interference by these agents with nearly all the mechanisms which normally form a co-ordinated ventilating system. Thus near critical doses may produce inhibition of the respiratory centres, impaired functioning of the respiratory muscles, increased secretions in the air-ways, and narrowing of the air-ways. The last two effects are of considerable importance if they offer an appreciable mechanical obstruction to air-flow in the lungs for, in addition to contributing to early respiratory failure, they may impede efforts to maintain adequate ventilation by artificial means.

Efforts to assess the degree of bronchial obstruction produced in humans by small doses of GB have led to somewhat vague results. Cooper and Maloney (1), estimating the effect by the Maximum Breathing Capacity test, reported a significant fall in MBC with dosages (Ct.) up to 6 mg.min./m³. Clements and his co-workers (2) however, as a result of measurements using the method of Otis and Proctor (3), concluded that the increase in air-way resistance with these low dosages was insignificant. Further work (4), in which the MBC test was employed, failed to show definite evidence of bronchial obstruction with larger dosages (up to 20 mg.min./m³), although there was a trend in the data towards a reduction in MBC. More recent work (5) also failed to gain evidence of bronchial obstruction.

The experimental results presented in this paper show that, in humans, the inhalation of more than 1.0 microgram/kg. of GB increases the lung resistance to flow, the effect being of a transient nature, and maximal immediately after absorption of the dose.

METHODS

1. Gassing

The use of Ct. as a measure of the dose was avoided for reasons of precision and safety. Measurements of the GB vapour concentrations and the volumes of air respired allowed more satisfactory estimates to be made of the doses in terms of the mass of GB absorbed. Doses from 0.6 - 3.0 micrograms/kg. were administered. The apparatus

S E C R E T

S E C R E T

described by Childs et al (6) was employed in most of the experiments. The subject inhaled orally from a two-way valve system, the volume of the expired air being measured with a direct-indicating spirometer. A two-way tap connected the subject either to fresh air or to a 100 m³ chamber in which the GB concentration was maintained. The retention of GB vapour inhaled orally has been estimated to be approximately 85% (7), and a correction by this factor was made in calculating the dose. In some experiments, the device shown in Fig. 3 was used. To charge the system, the main tube A of about 500 cc. in volume was evacuated to about 0.01 mm. Hg. The required dose of pure GB was dropped into a side vapouriser tube B using a recent development of a microburette previously described (8). The space volume of the vapouriser tube was as small as possible (about 1.0 cc.) and the tube was enclosed by a "basket" heater of nichrome wires capable of dissipating 30 - 40 watts. The tap between tubes A and B was opened and the heater switched on. The internal surface temperature of the vapouriser tube rose to about 60°C. within a few seconds, and the GB had evaporated in this time. Dry air was then admitted by opening tap T₁. The subject, wearing a nose-clip, exhaled fully and then inhaled orally from tube C, taps T₂ and T₃ having been opened. The inhaled volume was usually more than two litres, and tests with a mechanical system pumping this volume of air through the apparatus gave recoveries of 85 - 100% of the amount of GB evaporated.

2. Lung Resistance

Changes in lung resistance to air-flow were estimated using a resistance-interrupter method (9), and the electronic instrument already described (10). Values of air-way resistance for each individual are expressed in terms of the control value obtained before the experiment.

RESULTS

With few exceptions in more than sixty subjects, the air-way resistance to flow increased appreciably after the inhalation of the GB vapour. Table 1 shows the results of experiments in which, with the exception of the first group, tests were carried out 10 - 15 minutes after gassing. The mean values appear to show some correlation between dose and increase in lung resistance. The smallest doses (first group) in these experiments, however, produced effects which seemed disproportionately large, but, as will be shown, this apparent discrepancy was probably due to the difference in timing of the tests. In most cases, inhibition of the blood Ch E was estimated, the results being included in Table 1.

S E C R E T

S E C R E T

TABLE 1

THE INCREASE IN LUNG AIR-WAY RESISTANCE FOLLOWING EXPOSURE TO GB VAPOUR

Subject	Conc. (mg./m ³)	Dose (μg./kg.)	Time (min. after gassing)	Blood ChE depression (%)	Lung Resistance (x normal)
All	2.5	0.75	1	0	1.40
Bre		0.84		0	1.48
Pin.		1.08		0	1.56
Cra		0.58		0	1.19
Sco		0.65		7	1.25
Mean		0.78		0	1.37
Chap	4.9	1.17	10	4	1.40
Sh		1.35		7	1.35
McK		1.29		11	1.56
Ca		1.20		16	1.25
Ell		2.70		37	1.75
Mean		1.54		15	1.48
Ae	5.0	1.40	15	10	1.00
Et		1.35		6	1.08
Va		1.45		15	1.00
Mean		1.45		10	1.03
Je	7.5	1.34	10	14	1.26
Fe		2.06		23	1.33
Th		2.06		23	1.30
Car		2.10		17	1.09
San		1.30		12	1.31
Mean		1.77		18	1.26
Ro	10	2.07	10	-	1.40
McD		2.10		-	1.50
Woo		2.10		-	1.50
Wat		2.07		-	2.10
Mean		2.09			1.62
She	10	2.96	10	38	2.12
Bur		2.92		42	1.67
Dil		3.01		32	1.97
Bee		2.60		27	1.58
Kin.		2.87		46	1.50
Mean		2.87		37	1.77

Table 2 shows the results of further experiments in which estimates of air-way resistance were made at various times after gassing.

S E C R E T

S E C R E T

TABLE 2

The variation of lung air-way resistance with time after inhalation of CB vapour

Subject	Concn. inhaled ($\mu\text{g./m}^3$)	Volume respired (litres)	Dose absorbed ($\mu\text{g./kg.}$)	Lung Air-way Resistance (\times normal)							
				1	2	5	7	10	15	30	1 hr.
Coa	5	1.0	0.83	-	1.65	1.25	-	-	1.10	1.00	-
Jon	10	0.67	1.40	-	1.40	1.40	-	-	1.06	1.07	-
Le	20	1.50	1.63	-	-	1.22	-	-	1.12	-	-
St.	20	1.30	1.36	-	-	1.08	-	-	1.00	-	-
La	20	1.40	1.18	-	-	0.95	-	-	1.06	-	-
Iah	20	1.47	1.65	-	-	1.35	-	-	1.40	-	-
Lo	20	1.60	1.57	-	-	1.28	-	-	1.23	-	-
Cur	30	2.30	-	-	-	1.20	-	-	0.93	-	-
Far	30	2.10	-	-	-	1.60	1.23	1.35	1.18	-	-
She	10	2.0	2.96	-	-	-	-	-	2.12	-	-
Bur	20	2.92	-	-	-	-	-	-	1.67	-	-
Dil	20	3.01	-	-	-	-	-	-	1.58	-	-
Bee	20	2.60	-	-	-	-	-	-	1.97	-	-
Kin.	20	2.87	-	-	-	-	-	-	1.50	-	-
Edw	13.7	4	0.78	1.30	-	1.26	-	-	1.24	-	-
Trot	13.1	1.2	2.40	2.10	-	1.56	-	-	1.40	-	-
Smi	15	6.8	1.56	1.75	-	1.42	-	-	1.46	-	-
Coo	15	7.3	1.70	2.02	-	1.35	-	-	1.23	-	-
Hay	15	7.0	1.66	1.78	-	1.36	-	-	1.28	-	-
Dic	15	7.2	1.65	2.11	-	1.43	-	-	1.50	-	-

S E C R E T

0.95

1.06

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

-

SECRET

TABLE 2 (contd.)

Subject	Concn. Inhaled (mg./m ³)	Volume respired (litres)	Dose absorbed (μ g./kg.)	Lung air-way Res.-stance (x normal)								
				1	2	5	7	10	15	30	1 hr.	
Hyn	22	8	2.0	2.60	2.10	1.66	1.55	-	1.14	1.27	-	1.56
Lew		8	2.1	1.48	-	1.06	-	1.00	1.17	-	1.06	-
Woc		8	1.95	1.75	-	1.60	-	1.40	1.30	-	1.14	0.95
Gal	12	2.9	1.90	1.70	-	-	-	1.43	-	1.23	-	1.30
Hina	12	3.1	2.57	1.56	1.60	-	-	1.56	1.50	1.50	-	1.26
Coop	12	2.7	2.31	1.85	1.83	-	-	1.62	-	1.55	-	1.22
												0.95
ELL	30	3.2	1.5	-	4.40	-	-	-	2.82	-	1.61	1.60
Gre		3.4	1.34	-	2.16	-	-	-	2.10	-	1.95	1.64
Prie		3.0	1.27	-	1.84	-	-	-	1.56	-	1.33	1.33
Kah		3.0	1.40	-	2.14	-	-	-	1.59	-	1.19	1.23
Gos	35	3.0	1.49	3.10	-	2.07	-	1.80	-	1.80	2.10	1.40
Fen		2.8	1.41	2.75	-	2.40	-	1.94	-	2.00	1.82	1.35
Wid		3.4	1.74	3.00	-	2.20	-	1.86	-	1.33	1.33	1.25
Sta	40	3.0	1.50	2.20	-	-	-	2.00	-	2.00	1.52	1.40
Ken		3.0	1.35	4.09	-	-	-	1.88	-	1.57	1.18	1.28
Will		3.0	1.40	2.62	2.49	-	-	1.75	-	1.70	-	1.57

SECRET

S E C R E T

In nearly every case, the estimated air-way resistance increased to a peak value immediately after gassing. The resistance then diminished fairly rapidly, in some instances approaching the control value within 15 - 30 minutes. The time taken for the resistance value to return to normal depended on the degree of the initial effect. In a few cases an effect was measurable for more than 24 hours.

The subjective symptoms were described in a variety of ways. Most subjects, however, reported a transient "tightness in the throat" together with a more persistent "chestiness" sometimes described as being "a bit harder to breathe". These symptoms occurred during or immediately after inhaling the GB, and it was clear in some cases that the respiratory flow rates attainable were reduced. This was indicated by the lowered pressure differentials in the external resistance to flow. For a minute or so after gassing these subjects, in spite of repeated requests, were unable to exhale with flow rates as high as in the control tests.

The data in Table 2 were broadly analysed in two ways. Fig.1 shows the variation of average lung air-way resistance with time after gassing for selected groups in which the individuals received about the same dose. Fig.2 shows a similar plot of data for groups in which the individuals were exposed to the same concentration. Although complicated by the variation in average concentration from group to group, Fig.1 appears to indicate that the effect depended to some extent on the dose. Fig.2 shows that the magnitude of the effect was also dependent on the concentration of the vapour inhaled.

DISCUSSION

The experiments described have shown that an inhaled dose of GB vapour greater than 1.0 microgram/kg. and at a concentration higher than 5 mg./m³ produces a measurable increase in lung air-way resistance. Such a dose corresponds to a resting exposure to a dosage (Ct.) greater than 10 mg.min./m³ ($t < 2$ min.). The maximum increase in lung resistance occurred almost immediately after inhalation of the GB. This finding contrasts with that of Cooper and Maloney (1) whose experiments indicated that the maximum degree of air-way obstruction occurred from 15 minutes to 2 hours after gassing.

In a few cases, the lung resistance to flow increased for some minutes to as much as three to four times the normal value without causing any significant respiratory embarrassment. The respiratory reserve is normally large, and Comroe (11) has already pointed out that a substantial resistance to air-flow in the lungs can be tolerated in otherwise normal subjects. With larger doses of GB however, other factors which reduce the available respiratory effort would tend to make an air-way obstruction less tolerable.

The symptom described by the subjects as "chest tightness" or "a bit harder to breathe" seemed to follow the air-way obstruction in degree. Most subjects said that the effect was most noticeable immediately after gassing, and that it had disappeared or was considerably diminished after 15 - 30 minutes. Nevertheless, some subjects subsequently claimed to have experienced occasional "chestiness" over 24 hours,

S E C R E T

air-way resistance coefficient had increased. Repeated tests on a few normal subjects showed air-way resistance to be constant within the limits of error for periods of months, and it is possible that the increase in air-way resistance may be due to a change in lung resistance requiring even greater respiratory effort.

The increase in air-way resistance appeared to depend on the dose and concentration of inhaled GB vapour. Limited experiments were carried out at a single inhaled concentration, the apparatus shown in Fig.3 being employed. Concentrations of 0.4 micrograms/kg. were administered to five subjects as a single breath, the time taken for inspiration and expiration being from 5 - 8 seconds. The average value of the peak air-way resistance expressed in terms of the normal resistance was 2.55 (range 1.98 - 3.90). The increase was therefore slightly smaller than that found with concentrations from 30 - 40 mg./m³ and similar doses. The concentration of GB in the apparatus was about 200 mg./m³ but it was undoubtedly much reduced when mixed with the volume of air inhaled (more than 2 litres).

The transient nature of the air-way obstruction seemed to indicate a local effect followed by circulatory clearance of the active agent. Blood clearance of various agents has been shown (12), (13) to be a first order process, and the reduction of air-way resistance after gassing roughly followed such a relation with time. It is therefore of some interest, especially from the aspect of extrapolation from the present results, to consider the situation where the active agent is appearing in tissues at a constant rate and is being removed at a rate proportional to the amount of agent present. Suppose the rate of appearance of the active agent is aCV, where

$a = \text{constant}$,

$C = \text{concentration of inhaled vapour}$,

$V = \text{volume of air respired in unit time}$, and let

$k = \text{exponential clearance constant}$, and

$E = \text{amount of agent present in the tissues at the time } t$.

$$\text{Then} \quad \frac{dE}{dt} = aCV - kE \quad \dots \quad (1)$$

$$\text{i.e.} \quad E = \frac{aCV}{k} (1 - e^{-kt}) \quad \dots \quad (2)$$

or, if the dose absorbed in a time t is represented by D ,

$$E = \frac{aD}{kt} (1 - e^{-kt}) \quad \dots \quad (3)$$

These relations are represented graphically in Figs.4 and 5. In Fig.4 the relative amount of active agent in the tissues (E) is related to the time of exposure (i.e. dose), the inhaled concentration and the respiratory volume being assumed constant. In Fig.5 the total dose is

S E C R E T

S E C R E T

constant and E is related to the rate of absorption of the dose (i.e. to the concentration inhaled).

It is seen (Fig.4) that at a constant inhaled concentration, E rises with increasing time of exposure, but at a progressively decreasing rate. Fig.5 indicates that for a given dose, E increases at first as the inhaled concentration rises, but then approaches a value which remains almost constant for further increases in concentration. These qualitative results, although based on an elementary conception of the mechanism involved, do not conflict with the present experimental data, and will be examined in further experiments.

(Sgd). H. Cullumbine,

Head, Physiology Section.

(Sgd). E.A. Perron,

Supt., Research Division.

NA/JWE/GC.

S E C R E T

S E C R E T

References

1. Cooper D.Y., J.V. Maloney.	Medical Laboratories Research Report 82.	1951
2. Clements J.A., J.C. Moore, R.P. Johnson, J. Lymott.	M.L.R.R. 122.	1952
3. Otis A.B., D.F. Proctor.	Am.J.Physiol. <u>152</u> , 106.	1948
4. Gullumbine H., S. Callaway.	P.T.P.321.	1952
5. Freeman G., J.A. Clements, et al.	M.L.R.R.148.	1953
6. Childs A.F., N. Creasey, D.R. Davies, G. Kirkham.	P.T.P.357.	1953
7. Oberst F.W., W.S. Koon, J.W. Crook.	M.L.R.R.143.	1953
8. Ainsworth, M.	Porton Report 2651.	1944
9. Ainsworth M., J.W. Eveleigh.	P.T.P.320.	1952
10. Ainsworth M., J.W. Eveleigh.	P.T.P.331.	1953
11. Comroe, J.H.	Progress Report, University of Pennsylvania. Contract DA-18-108 CML 2212.	Aug. 1951
12. Kety, S.S.	Am.Heart.J. <u>38</u> , 321.	1949
13. Cruickshank J.D., Ainsworth M.	P.T.P.345.	1953

S E C R E T

FIG. I. VARIATION IN AIR-WAY RESISTANCE AFTER INHALATION
OF G.B. VAPOUR.

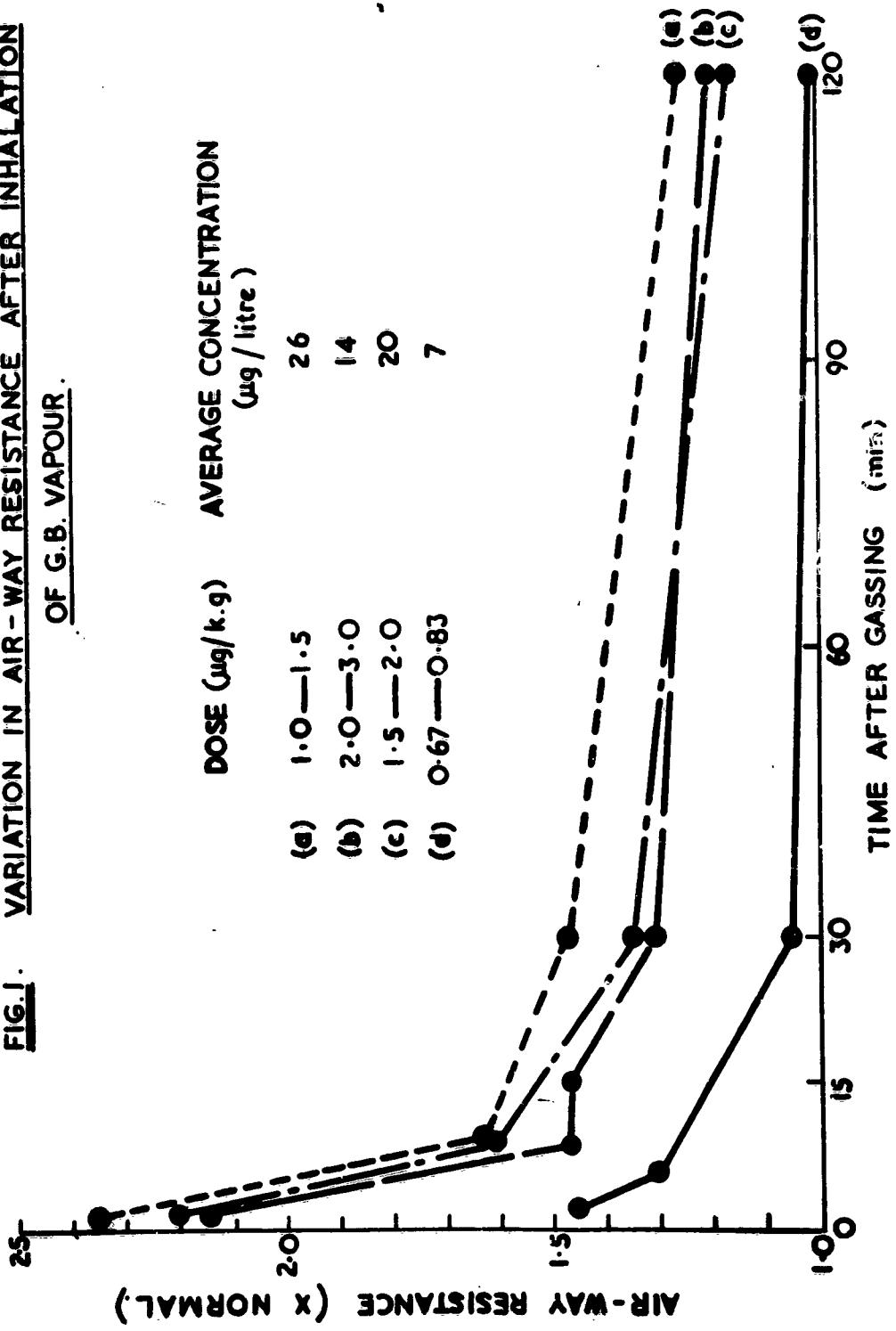


FIG. 2. VARIATION IN AIR-WAY RESISTANCE
AFTER INHALATION OF C.S. VAPORIS

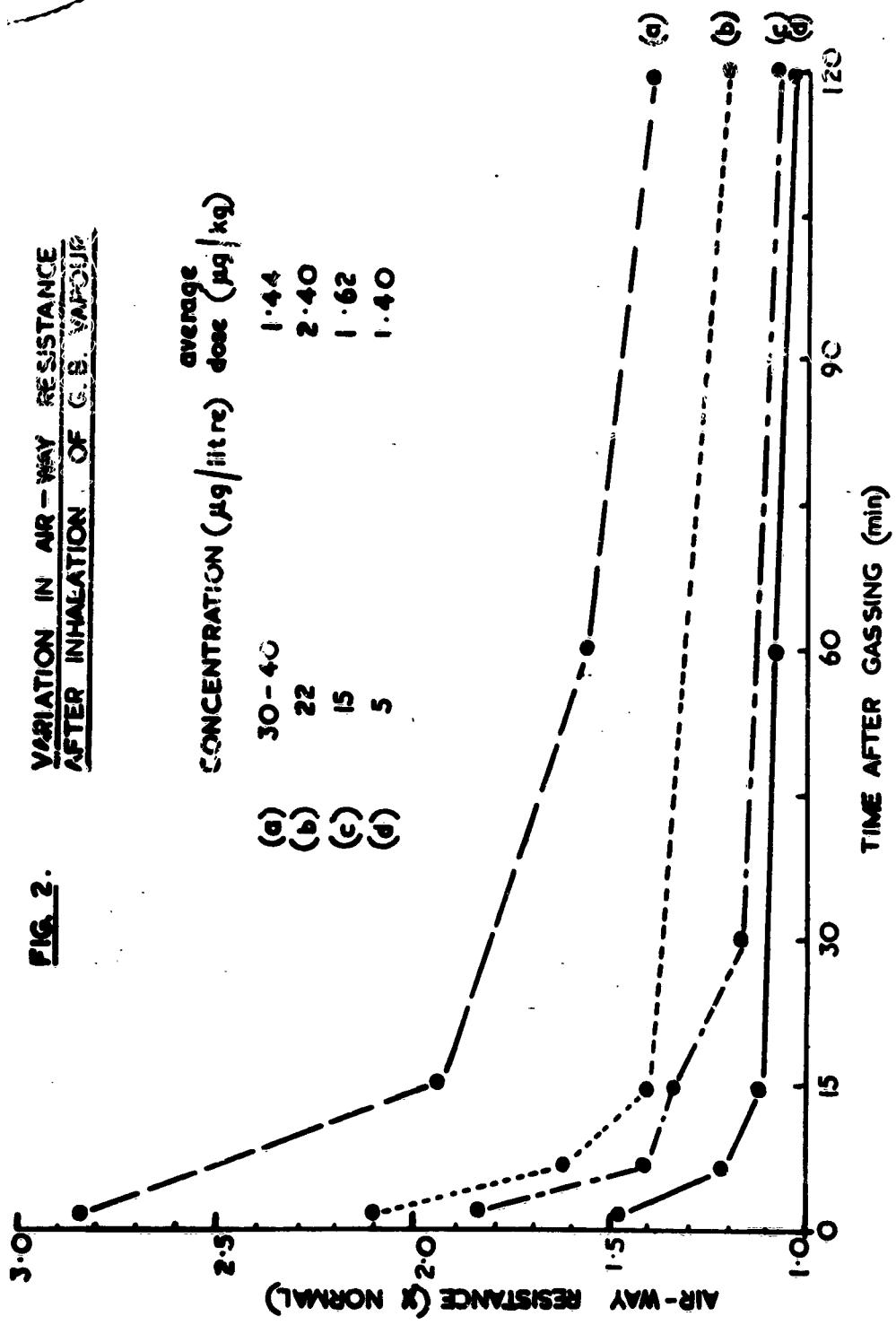
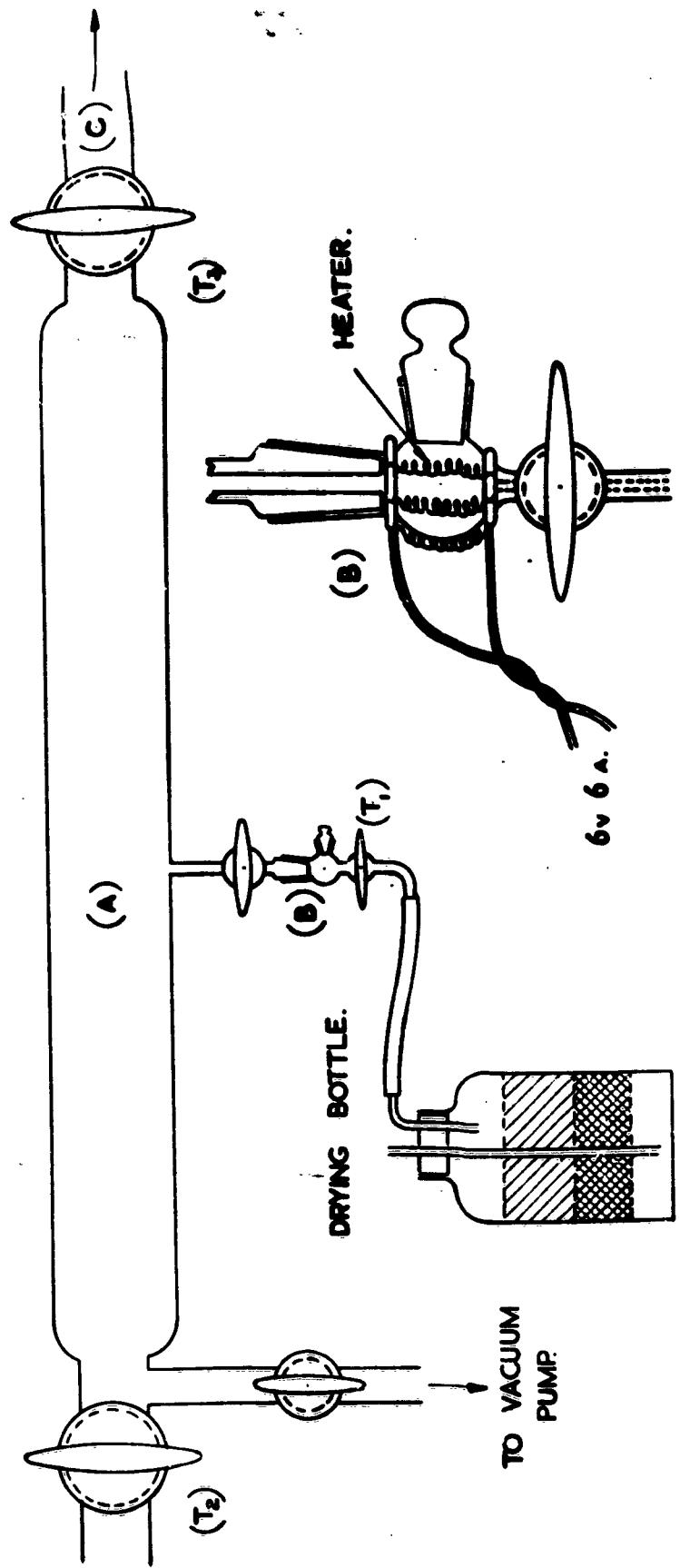


FIG. 3. DEVICE FOR ADMINISTERING KNOWN DOSES OF GB VAPOUR.



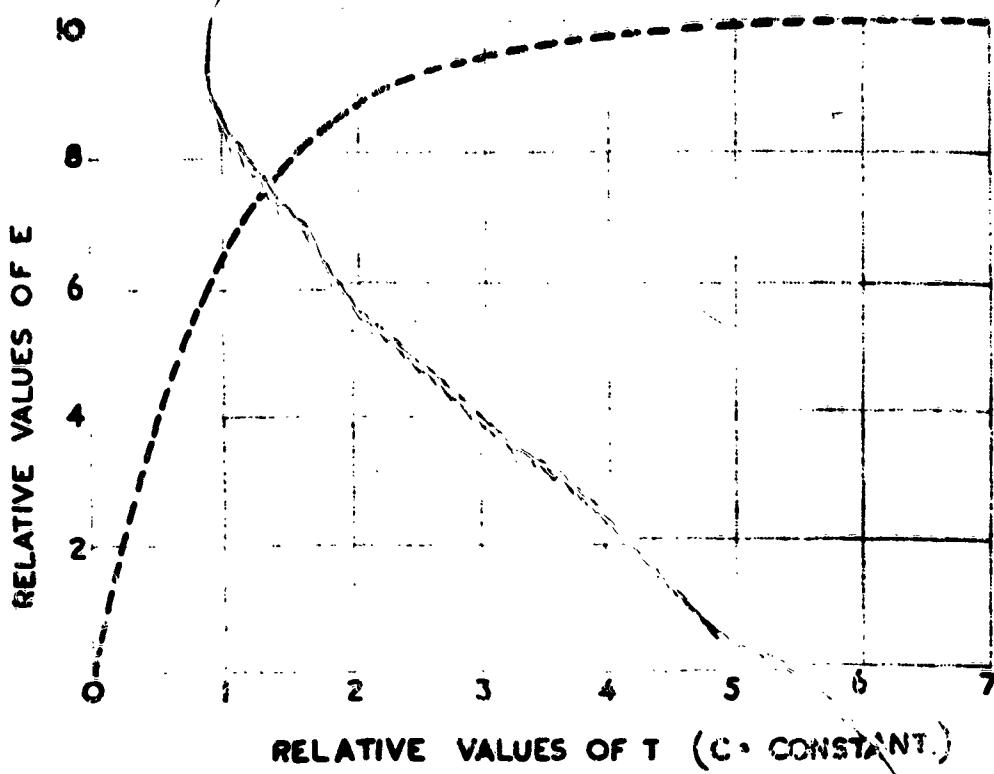


FIG. 4.

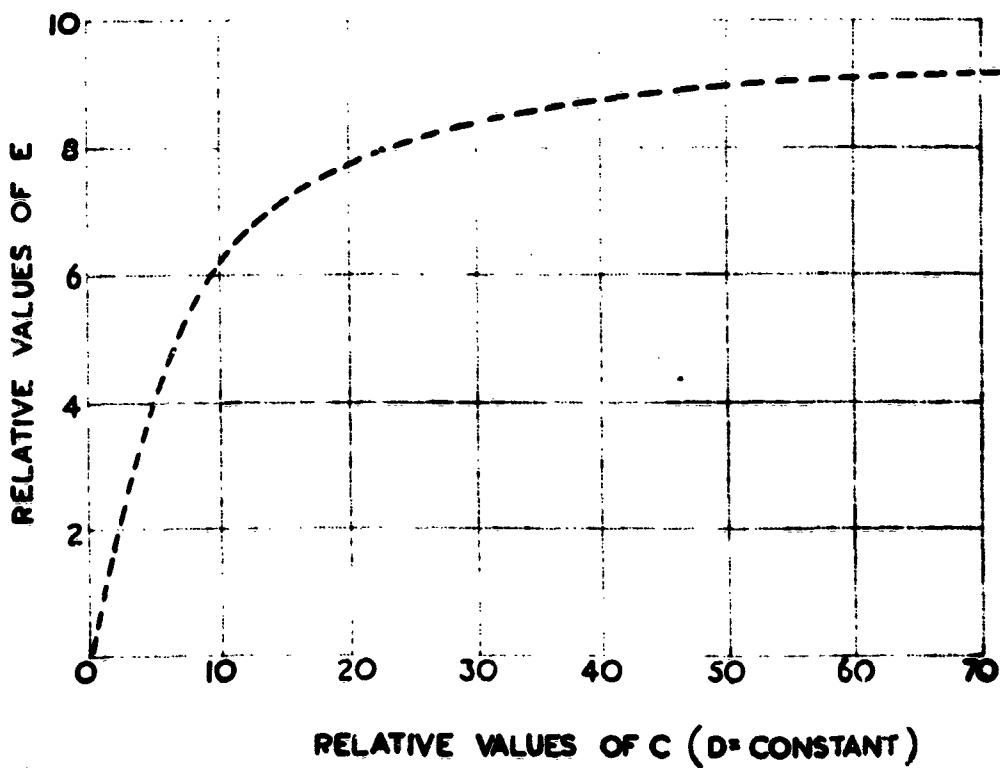


FIG. 5.

P.T.P.371

CIRCULATION

INTERNAL

Copy No.

1	P.D.S.R.(D)
2	D.C.D.R.D.
3	A.D./C.D.R.D.
86 - 115	C.S., C.D.E.E.
4 - 5	C.D.R. Branches
6 - 7	C.S., M.O.S. Nancekuke
8	D.P.B.R.
9	R.D.Arm.2.
10 - 11	T.P.A.3/T.I.B.
12 - 26	S.A.C. (for Biology Committee, C.D.A.B.)
27	File

Chemical Defence Advisory Board

28	Professor Sir Rudolph Peters
29	Professor G.R. Cameron
30	J. Davidson Pratt Esq.
31	Professor H.J. Emeleus
32	Sir Paul Fildes
33	Professor J.H. Gaddum
34	Professor H.W. Melville
35	Professor Sir Robert) Robinson
36	Professor R.H.S. Thompson

Biology Committee

37	Dr. J.M. Barnes
38	Professor I. de Burgh Daly
39	Dr. G.S. Dawes
40	Dr. Malcolm Dixon
41	Dr. Bentley Purchase
42	Professor A. Wilson

EXTERNAL

Copy No.

British Joint Services Mission

43 - 54 D.C. Evans Esq.

War Office

55 S.W.V.1.(b)

OVERSEAS

(through T.P.A.3/T.I.B.)

Australia

56 - 58	Defence Res.Laboratories
59	Senior Rep., Dept. of Supply
60	Army Branch Representative
61	R.A.A.F. (Tech. Section)

Canada

62 - 63	Chairman, Defence Research Board
64 - 65	Defence Res.Labs., Ottawa
66	Suffield Exptl. Station

U.S.A.

67 - 84 Reading Committee



Information Centre
National Archives
[dstl] Porton Down,
Salisbury
Wiltshire
SP5 2PF
Telephone 01258
03550 33783
Fax 01258 621570

Defense Technical Information Center (DTIC)
8725 John J. Kingman Road, Suit 0944
Fort Belvoir, VA 22060-6218
U.S.A.

AD#: AD014839

Date of Search: 5 March 2009

Record Summary: WO 189/705

Title: Air-Way Resistance Changes in Men Exposed to GB Vapour

Availability Open Document, Open Description, Normal Closure before FOI Act: 30 years

Former reference (Department) PTP 371

Held by: The National Archives, Kew

This document is now available at the National Archives, Kew, Surrey, United Kingdom.

DTIC has checked the National Archives Catalogue website (<http://www.nationalarchives.gov.uk>) and found the document is available and releasable to the public.

Access to UK public records is governed by statute, namely the Public Records Act, 1958, and the Public Records Act, 1967.

The document has been released under the 30 year rule.

(The vast majority of records selected for permanent preservation are made available to the public when they are 30 years old. This is commonly referred to as the 30 year rule and was established by the Public Records Act of 1967).

This document may be treated as UNLIMITED.